Annual Report 2008

Department of Chemical Engineering



DEPARTMENT OF CHEMICAL ENGINEERING, NTNU

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Professor Heinz Preisig Professor Hallvard Svendsen Associate Professor Hilde J. Venvik Senior Engineer Berit Borthen PhD-Candidate Ellen Marie Flaten Student Stine T. Roset Student Rebecca Williams

Staff

Academic staff, see the individual research groups:

Technical and administrative staff:

Head of Administration Tom Helmersen

Administrative staff:

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Technical staff:

Senior Engineer Berit Borthen Engineer Harry Brun Engineer Arne Fossum Engineer Odd Ivar Hovin Principal Engineer Signe Håkonsen Principal Engineer Iva Králová Executive Officer Tove Barø Executive Officer Gerd Sandvik

Principal Engineer Caterina Lesaint Principal Engineer Andrea A. Mingot Principal Engineer Jan Morten Roel Principal Engineer Cecilie M. Selsbak Engineer Frode Sundseth Senior Engineer Bodhild Øvrevoll Senior Engineer Asbjørn Øye

Department of Chemical Engineering Head: Sigurd Skogestad

Advisory

committee

Administration and technical staff

Catalysis group Colloid and polymer chemistry group Enviromental engineering and reactor technology group Paper and fibre technology group

> Process systems engineering group

COVER PAGE (Illustration by Alexey Voronov and Vladislav Kalitka)

In situ microreactor for combined X-ray diffraction, X-ray absorption spectroscopy and Raman spectroscopy: The reactor is used at the European Synchrotron Radiation Facility to study the Fischer-Tropsch synthesis catalyst at realistic working conditions.

DEPARTMENT OF CHEMICAL ENGINEERING, NTNU

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CHAPTER 1: INTRODUCTION

Introduction to Annual Report 2008

By Sigurd Skogestad (Head of Department)



This report aims at summing up the activity in 2008 as well as providing more general information about the department. We hope that you find it useful!

In 2008 a shift occurred towards a more international department. First, we admitted our first 9 students in the *international master program* in chemical engineering. All teaching in the 4th ond 5th year is now offered in English and the international master students follow the same courses as the ordinary Norwegian students in their two years of study.

Second, our PhD population is changing from largely Norwegian towards being truly international. The 38 new PhD students in 2008 (a record by a large margin!) are from the following 15 countries:

Norway	6
India	6
Iran	4
China	3
Czech Republic	3
Pakistan	3
Russia	3
Greece	2
Romania	2
Albania	1
Argentina	1
Bangla Desh	1
Netherlands	1
Slovenia	1
Tanzania	1

In 2008, the research output, as expressed by the number of graduated PhD candidates and publications, was good, although not quite as high as in the record breaking year 2007. The number of credited publications in international journals was 84 (2008), whereas it in the five previous years was 46 (2003), 48 (2004), 76 (2005), 74 (2006) and 108 (2007).

Furthermore, 13 PhD candidates graduated in 2008, which is about the same as previous years. Since the avarage time to complete a PhD is about 4-5 years (including leaves of absense), we can infer from the Table on the next page that the PhD production is likely to to drop slighly the new two years, but will then increase sharply because of the record number of 38 new PhD students in 2009, of which many are

funded by the new GASSMAKS program of the Norwegian Research Council (NFR).

A problem for the Department over the last 7 years or so has been the low number of students in the *Chemical Engineeringy and Biotechnology* program. Partly in response to this situation, we have started a 2year International Master program in Chemical Engineering. We presently have 9 students and are expecting to expand to 15-20 new students each year.

In terms of plans and faculty recruitment, we have followed quite closely the Departments strategic plan from 2003. In January 2008, Nadav ("Nadi") Skjøndal-Bar started as Associate Professor (*Førsteamanuensis*) in the field of systems biology. This is cooperation with similar activities in the Departments of Biotechnology and Biology. The Department has now completed a major replacement of its academic staff with 15 (of 19) new fulltime faculty members being hired since 1997.

The Department is located in Chemistry buildings K4 and K5, experimentall halls C and D (which could be considered building K6) and in the PFI-building across the street. Our chemistry buildings date back to 1957/58 (K5/Exp. Halls) and 1965 (K4). Due to inadequate ventilation in K4, some of the labs in the basement and ground floor have not been in use since the paper group moved out in 1998, and the two upper floors have been largely empty since the electrochemistrry group moved out in late 2005. A complete renovation of K4 will take place in 2009, and is expected to be finished early 2010. The paper machine in the basement, which was officially opened by King Olav in 1967, will be removed to make place for new labs for the expanding CO2-removal activity. The funding for the renovation comes from the Government Financial Crisis Package (120 mill. NOK) and from the university (25 mill. NOK). This funding will also be used to start renovating the experimental halls, but a complete renovation will require additional money.

In 2008 the Department was evaluated by the Research Coincil as part of an evaluation of all Chemistry Research in Norway. The evaulation was very postive for the Department, and the committee writes: "The Committee was very favourably impressed by the research in this department, and further believes that the close interactions with SINTEF and other research entities are highly beneficial. The Committee can therefore make no major general recommendations for changes, other than that the Committee is in favour of maintaining current directions and plans."

FACTS ABOUT THE DEPARTMENT OF CHEMICAL ENGINEERING

The Department of Chemical Engineering is located at the <u>Gløshaugen campus</u> of the Norwegian University of Science and Technology (NTNU) in Trondheim. NTNU is the only university in Norway that awards engineering degrees in all areas.

The Department offers a 5 year program leading to the degree of *sivilingeniør* (M.Sc.) in chemical engineering. Most of the students start at NTNU in their first year, but about 10 to 20% enter in the fourth year based on a 3-year engineering Bachelor degree. On top of this we offer a 3 year doctoral program leading to a Ph.D. degree in chemical engineering. In addition, we offer a 2 year International Master Program in Chemical Engineering.

The Department can trace its roots back to 1910 when the Norwegian Institute of Technology (NTH) started up in Trondheim with engineering chemistry as one of the seven majors. After the Second World War, three applied Departments were formed, namely pulp and paper chemistry (*treforedlingskjemi*, 1946), chemical engineering (*kjemiteknikk*, 1949) and industrial chemistry (*industriell kjemi*, 1950). These merged in 1999 to the present Department of chemical engineering (*kjemisk prosessteknologi*).

The objectives of the Department are:

- 1. *Education*. Offer a Master Degree in Chemical Engineering which is internationally recognized and makes the candidates attractive on the labour marked.
- 2. *Research*. Research shall be on an international level, and in some areas internationally leading.
- The Department shall be attractive in order to recruit the best candidates, including academic faculty, PhD students and undergraduate students. The social environment shall be very good so that everyone feels welcome.

The permanent staff in 2008 included

- 17 technical/administrative
- 19 academic, incl.12 Professors and 7 Associate Professors (*Førsteamanuensis*)

The non-permanent staff in 2008 included

- 4 technical
- 5 Adjunct Professors (*Professor II*) (20% *position*)
- 98 PhD students
- 28 Post.docs and researchers

The Department also houses 8 Professor emeritus and 4 visitors, in addition to a large SINTEF group.

Student production

Year	MSc	PhD
1994	60	11
1995	79	2
1996	57	5
1997	67	9
1998	46	13
1999	81	8
2000	69	10
2001	$18^{(*)}$	11
2002	75	12
2003	44	7
2004	30	10
2005	25	13
2006	19	15
2007	31	15
2008	31	13
MSc studen	n from 4.5 to 5 ye	program.
5 th year	33	
4 th vear		International Master)
3 rd year	26	
	tudents (exchan	ge students not included)
2003		20
2004		10
2005		9
2006		18
2007		15
2008		38
	ocs/Scientists	
2006		10
2007		25
2008		28

CHAPTER 2: RESEARCH



CATALYSIS GROUP

Academic staff

Professor Anders Holmen Professor Edd A. Blekkan Professor De Chen Professor Magnus Rønning Associate professor Hilde J. Venvik Adjunct professor Kjell Moljord Adjunct professor Erling Rytter

Post docs

Anna Maria Lind (until 31.08.08) Espen Standal Wangen (from 05.05.08) Estelle Vanhaecke (from 18.02.08) Hongmin Wang (from 27.02.08) Santhosh Kumar Matam (until 04.06.08) Tiejun Zhao

PhD candidates

Alexej Voronov (from 30.04.08) Anh Hoang Dam Astrid Lervik Mejdell Bjørn Christian Enger (until 31.10.08) Eleni Patanou (from 03.03.08) Fan Huang (from 31.01.08) Fatemeh Haver Hamidreza Bakhtiary Hassan Jamil Dar (from 01.08.08) Hilde Meland (until 30.05.08) Ilya Gorelkin (from 28.07.08) Ingvild Tronstad (from 01.10.08) Jia Yang (from 01.08.08) Juan Maria Schena (until 29.02.08) Li He Miroslav Surma (from 12.08.08) Navaneethan Mutuswamy (from 15.05.08) Nikolaos Tsakoumis (from 03.03.08) Nina Hammer (until 15.11.08) Oana Mihai (from 02.05.08) Paul Radstake (14.08.08) Saima Sultana Kazi Sara Boullosa Eiras Šárka Zárubová (from 25.08.08) Shreyas Pandurang Rane (from 02.01.08) Silje Fosse Håkonsen (until 15.06.08) Tayyaba Noor (from 01.04.08) Xuven Kim Phan

Guests

Fernando Bimbela (PhD student from Zaragoza, Spain) Matteo Lualdi (PhD student from KTH, Sweden) Sara Lögdberg (PhD student from KTH, Sweden) Yian Zhu (Guest rechearcher from East China Univ. of Science, China) Johan den Breien (PhD student from Utrecht Univesity, The Netherlands) Romas Skudas (PhD student from Johannes Gutenberg University, Mainz, Germany

The Organization

The research and teaching in catalysis, petrochemistry and related subjects (including surface science, adsorption and physical studies of porous materials, reaction kinetics and process engineering) is organised in the Catalysis Group, a joint effort where <u>NTNU</u>, and the research company <u>SINTEF</u> share laboratories and equipment.

Catalysis and Petrochemistry



Personnel from the two organisations work together and participate in teaching and research. About 10-15 students graduate each year (M.Sc.). The group participates extensively in international networks, research programs etc., and cooperates closely with a number of universities and research groups inside and outside the EU.

The group and the laboratories

At present the group comprises about 50 people: 5 Professors, 2 Adjunct professors, about 10 fulltime research scientists holding Ph.D's, 4 Post.doc's and about 25 Ph.D students. The laboratories and equipment include a large number of microreactors for catalyst studies, several small pilot plants, all the necessary equipment for catalyst and material characterization (chemisorption, physical adsorption, Temperature Programmed techniques (TPR, TPD, thermal analysis), XPS, Auger spectroscopy, STM, FTIR and others). Recently, in situ IR/Raman and the TEOM-technique (Tapered Element Oscillating Microbalance) have been introduced in the laboratory, and we were the first group in Europe to utilize the TEOM technique in catalyst studies. Cooperation with the Departments of Physics (TEM and surface science), and Materials Science and Engineering, the other groups at the department of Chemical Engineering (all aspects of chemical and process engineering, particularly reactor engineering and colloid and polymer chemistry) and other departments ensures a wide scope and a high quality of the work. The research is funded by the Norwegian Research Council and by industry and spans from fundamental studies of

ideal surfaces to studies of real catalysts to process development work in small pilot plans.

The projects

A description of the Group as well as further details of all the projects, are given in our Annual Report.

Natural Gas Conversion

Natural gas is an abundant hydrocarbon fuel and chemical feedstock, and utilizing this resource with minimum environmental impact is a major challenge to catalysis. It is the main goal of the present programme to study catalytic processes for conversion of natural gas to chemicals and fuels including hydrogen. The programme includes production of synthesis gas, Fischer-Tropsch synthesis, and dehydrogenation of C₂-C₄ alkanes. The work is carried out in close collaboration with Norwegian industry and SINTEF. The group also participates in a Centre for Research Innovation (SFI-inGAP) focusing on the use of natural gas.

Hydrogen Technology

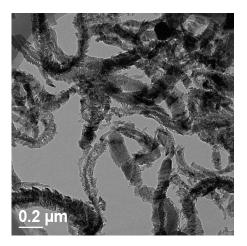
Particular attention is directed towards hydrogen technology: Catalysis is important in the production of hydrogen from hydrocarbons. Natural gas in an important source of hydrogen, and research is thus linked to syngas issues. In addition, the conversion of "transportable" hydrogen carriers such as propane, methanol and (bio) ethanol is studied. Of particular relevance is the integration of CO₂ separation technologies in hydrogen production processes, and this is targeted through sorption enhanced reactions and membrane reactors (see below). The group is also involved in development of improved fuel cell catalysts based on carbon nanofibers (also below). Collaborations include SINTEF as well as Norwegian industry. Hydrogen technology is also part of the MIT-NTNU cooperation.

Design and Preparation of New Catalysts and Supports

The catalytically active material is the key to any catalytic process, and the preparation of these, highly specalized functional materials is an important industry. Understanding the preparation methods, and developing new techniques is therefore a central research area. This programme deals with new methods to prepare supports and catalysts such as flame spray pyrolysis and spray drying, as well as the preparation and use of structured, mesoporous supports. This work is done in collaboration with the Ugelstad laboratory and SINTEF.

Carbon Nanofibres

Carbon nanofibres (CNF) have several interesting properties such as high resistance to strong acids and bases, high electric conductivity (similar to graphite), relatively high surface area and high mechanical strength. These unique properties lead to a large number of applications, such as catalyst supports, selective sorption agents, energy storage, composite materials, nano-electric and nano-mechanical devices, as well as field emission devices. The programme includes synthesis of carbon nanofibres and nanotubes of different morphology and the use of CNF/CNT in applications such as heterogeneous catalysis, fuel cells and conversion and storage of energy. This is done in collaboration with other groups at NTNU, SINTEF and Norwegian Industry



Carbon Nanofibres (TEM image)

Novel reactor concepts and structured supports Emerging reactor technologies such as microstructured reactors and (catalytic) membrane reactors are being developed and tested. The use of structured supports such as monoliths and foams is being studied, particularly for short contact time reaction systems such as partial oxidation and oxidative dehydrogenation. The work on microstructured reactors, where channels micrometer dimensions (1-1000µm) and up-scaling by parallelization is applied to enable new properties/possibilities, is performed in collaboration with Forschungzentrum Karlsruhe in Germany. Membrane reaction concepts based on novel Pd thin film technology are being developed together with SINTEF, and a partnership with MIT, and StatoilHydro is directed towards the use of hightemperature proton-conducting membranes in hydrogen production with CO₂ capture.

Oil Refining

Upgrading of crude oil and oil fractions is an important subject of research, especially due to new environmental legislation demanding more efficient processes. The programme includes catalytic reforming, isomerization, hydrotreating/ hydrocracking and heavy oil upgrading. The work is carried out in close cooperation with SINTEF and the industry.

Biofuels

The research is focused on catalytic aspects of thermochemical conversion, such as syngas adjustment, residual hydrocarbon reforming and Fischer-Tropsch synthesis.



In the EU project "GreenSyngas" the Catalysis group works on reducing the methane and tar contents in the syngas from the gasifier in Güssing (Austria) where woodchips is the raw material.

Fundamental Studies in Heterogeneous Catalysis

Several experimental techniques are used to study the details of solid catalysts. We are working together with Department of Physics on the use of Transmission Electron Microscopy and Scanning Tunneling Microscopy. We focus on characterisation of catalysts at working conditions and for this purpose we are using the European Synchrotron Radiation Facility in Grenoble and together with the Ugelstad Laboratory we have recently purchased new facilities for IR and Raman spectroscopy. The TEOM (Tapered Element Oscillating Microbalance) is also a powerful technique for studying important phenomena like catalyst deactivation, diffusion in porous materials and adsorption, absorption and desorption. Photocatalysis is also introduced as an important topic in the group.

COLLOID- AND POLYMER CHEMISTRY GROUP (UGELSTAD LABORATORY)



Academic staff

Professor Johan Sjöblom Associate professor Wilhelm R. Glomm Associate professor Gisle Øye Adjunct professor Egil Gulbrandsen (until 30.09.08) Adjunct professor Per Stenius (until 14.09.08) Professor emeritus Arvid Berge Professor eneritus Preben C. Mørk

Scientists

Sebastien Simon Brian Grimes Cédric M. Lesaint Kristofer Paso

Post docs

Yanru Fan Sondre Volden

PhD candidates

Asal Amiri Umer Farooq Serkan Keleşoĝlu Erland Nordgård Jamil Ahmad (from 15.05.08) Ann-Mari Dahl Hanneseth (until 31.07.08) Simone Less (until 31.07.08) Agnethe Knudsen (from 15.08.08) Nild van der Tuuk Opedal (from 15.08.08) Anne Silset (until 19.09.08) Dorota Dudásová (until 30.09.08) Martin Smestad Foss Klodian Xhanari (from 15.01.08)

Technical staff

Signe Håkonsen Caterina Lesaint Iva Králová (from 15.08.08) Cecilie M. Selsbak (until 20.08.08) Bodhild Øvrevoll (until 31.07.08)

Overview:

The Ugelstad Laboratory was founded in honour of Professor John Ugelstad at the Norwegian University of Science and Technology in January 2002 (Department of Chemical Engineering). The laboratory specializes in surfactant chemistry and its technical applications, emulsions and emulsion technology, preparation of polymers and polymer particles and their technical applications, plasma chemical modification of surfaces and silica-based chemistry.

Applications include crude oil production and processing, pulp and paper, biomedicine, catalysis and materials science.

The main purpose is to raise the national level of colloidal science by establishing a modern educational, research and development laboratory within the field of colloid, polymer and surface chemistry. Diploma and Ph.D. studies are offered within these topics, often in close collaboration with industrial companies. The aim is to educate highly qualified candidates for industrial positions. In order to attract the best and most motivated students and researchers, the laboratory has invested in new and modern instrumentation. The laboratory also participates in international exchange programmes, and hosts internationally renowned guest researchers and lecturers.

The Ugelstad Laboratory is sponsored by industrial companies, the Research Council of Norway (NFR), research institutes and NTNU. All the members are annually invited to a presentation of the recent research activities at the laboratory. This is combined with the Ugelstad Lecture, where invited scientists lecture within the field of colloid, polymer and surface chemistry.

Research Activities:

In the following paragraphs, selected ongoing research programs for 2008 are briefly described. For a complete description of the research activities at the Ugelstad Laboratory, please visit our web page: http://www.chemeng.ntnu.no/research/polymer/ugelstadla b/

Improved Oil recovery by Low Salinity Waterflooding: Surface Chemistry and SCAL Studies (VISTA 2007 – 2010)

The project focuses on how the surface chemistry of reservoir surfaces can contribute to improved oil recovery by low salinity waterflooding. An important goal is to understand the interaction mechanisms between the solid surfaces and surface active components in crude oils as the extension of the electrical double layer varies.

Multiphase Flow Assurance Innovation Centre (FACE) – Centre for Research-Based Innovation (SFI) 2006-2014

An increasing fraction of hydrocarbon reserves are difficult or impossible to produce and process today mainly due to the complexity of the fluids. Production of these reserves will require new and innovative technologies. FACE will develop the knowledge base for the new predictive tools that will be essential in order to develop the new, innovative production solutions. It is expected that new SMB's will be generated based on knowledge and technology from the centre as well as development of new or improved products in existing companies.

The research is focused on transport and separation aspects of three thematic topics, i.e. heavy crude oils, dispersed systems (emulsions), and solid particulate suspensions (hydrates, wax, sand and fines). Fluid characterization is a central tool to describe complex fluids within the three thematic topics and a necessary input to hydrodynamic modeling. We will use existing laboratories to perform both smallscale and high-pressure, large-scale flow experiments in pipes and separators. These experiments will be accompanied by multidimensional model development and their verification.

Collaborative effort between NTNU, SINTEF, IFE and UiO.

An Integrated Approach to Interfacial/Surface Processes in Crude Oil Systems (NFR PETROMAKS)

In crude oil production and processing heterogeneous systems play an important role in the whole value chain, starting from oil recovery from the reservoir and ending with crude oil refining in the refineries. A rational definition of heterogeneous systems is systems with either interfaces or surfaces or both. A brief survey of such systems and processes reveal wettability of the mineral surfaces in the reservoir as extremely important in the recovery of oil, chemical composition of the interface between water and oil as important for the separation process (and droplet-droplet coalescence) with consequences for both oil and waste water quality, corrosion inhibition of steel surfaces to prevent corrosion to deteriorate pipelines and the contamination of solid surfaces to poison catalysts. A general problem including most of the sub-processes in the exploitation of crude oil is a proper lack of instrumentation and competence to undertake a proper characterization of surface structures and concentrations in order to improve the understanding of oil recovery and especially enhanced oil recovery, separation technology and maintenance of pipeline integrity and transport.

Treatment of Produced Water: Characterization and New Treatment Strategies. Petromaks program (NFR).

The research tasks in this program will contribute to the development of new and improved technology for a more efficient and cost effective treatment of produced water from offshore installations. A fundamental necessity of developing and designing any treatment scheme is the knowledge and understanding of the fluid to be treated. The work on treatment strategies will focus on two areas: treatment of suspended constituents and of dissolved/soluble constituents. A major factor in achieving a zero harmful discharge to sea is the removal of suspended solids and dispersed oil from produced water. Particle separation is a fundamental process in any treatment process for the production of high quality effluent from an aqueous stream. The removal of dissolved constituents in produced water is necessary within the zero harmful discharge network. One of the

research tasks in this proposal is to investigate the applicability of biological degradation of specific target compounds.

This programme ended in 2008 with the dissertation of 2 PhD students. The main outcome of the research tasks were identification of variables contributing to the stabilisation of solid suspensions and o/w emulsions. The fundamental understanding of the stabilisation mechanisms is valuable for ensuring reliability and optimisation of existing produced water treatment processes, as well as in developing existing and new treatment technologies.

Development of nanostructured microreactor-based reaction technologies for continuous in situ production of food and drug intermediates from renewable resources. Case studies of synthesis of gluconic acid and enantiopure components of natural flavors (Polish-Norwegian Research Fund) 2008-2010 The project aims at developing novel family of high performance technologies for the in situ production of valuable chemicals – natural intermediates for food and pharmaceuticals synthesized from renewable resources, e.g. cellulose materials. The development of effective, high throughput technologies for in situ (local) production of valuable chemical intermediates from renewable resources is one of the top priorities of EU for sustainable development.

Thermoresponsive Polymer Brushes on Nanoparticles and Surfaces (NFR FRINAT 2007-2010)

An interesting category of amphiphilic polymer systems are those who undergo phase transitions in response to environmental stimuli such as temperature and pH. These have been widely investigated for drug delivery, separations and diagnostics applications. The aim of this project is to develop an understanding - both mechanistic and applied - of different thermo-responsive polymers adsorbed to planar surfaces and particle substrates. Changes in structural and dynamic properties of the systems under various conditions will be examined.

This project is a collaborative effort between the Ugelstad laboratory and Dept. of Chemistry, UiO, with several international partners.

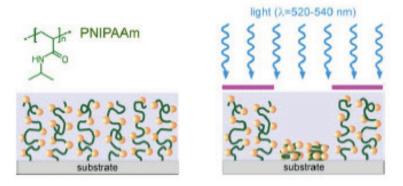


Figure 1: Schematic illustration of the "T-jump" modulated conformational transitions in surface-anchored PNIPAAm brushes. Gold nanoparticles will be loaded inside the brush and will attach to the NIPAAm monomers via H-bonds. Light having a wavelength close to the plasmon band of Au will be used to locally heat the particles, which in turn will cause collapse of the PNIPAAm chains. Through employment of a mask placed between the light course and the polymer, the PNIPAm collapse can be spatially modulated.

Structure, behaviour and reactivity of tetrameric naphthenic acids (ARN) in bulk and at w/o interfaces 2005-2008

The naphthenate R&D group at Statoil has done very systematic work during the past years to identify the structure of the naphthenic acid being the most active in forming metalnaphthenate deposits, which is a severe obstacle in processing of acid crudes. The results from the Statoil research in this field are pioneering, and have lead to the discovery of the socalled ARN naphthenic acid, which represents an acid family of C80 tetramers. The results from the Statoil discovery have recently been published, and one can foresee a heavy international scientific follow-up in this area in the years to come.

The objective of this programme is to focus on combining the efforts of the Statoil Naphthenate R&D Group and from our recently completed VISTA project. In the new VISTA programme, we are going to undertake a fundamental study of the ARN family of naphthenic acids with regard to clarify the structure(s), the physico-chemical properties, the interfacial activity and reactivity, selectivity in reaction patterns with multivalent cations, filmforming properties, etc.

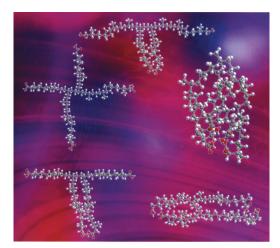


Figure 2: Archeal C_{80} isoprenoid tetraacids responsible for naphthenate deposition in crude oil processing.



Official opening of the new laboratory in 3rd floor, Chemistry building 5 (K5) by Head of Department Sigurd Skogestad and Professor Johan Sjøblom.

PROCESS SYSTEMS ENGINEERING GROUP

Academic staff

Professor Sigurd Skogestad Professor Terje Hertzberg Professor Heinz A. Preisig Associate professor Tore Haug-Warberg Associate professor Nadi S. Bar Professor emeritus Terje Hertzberg

Post docs

Sridharakumar Narasimhan (until 26.06.08) Mohammad Samsuzzoha (from 29.09.08)

PhD candidates

Andreas Linhart Bjørn Tore Løvfall (until 31.07.08) Elvira Marie B. Aske Heidi Sivertsen (until 19.12.08) Henrik Manum Håkon Dahl-Olsen Ivan Dones Jens Petter Strandberg Johannes Jäschke Jørgen Bauck Jensen (until 16.05.08) Jørgen Skancke (from 01.09.08) Magnus Glosli Jacobsen Maryam Ghadrdan (from 16.07.08) Mehdi Panahi (from 07.01.08) Olaf Trygve Berglihn Ramprasad Yelchuru (from 16.01.08)

Guests

Jakub Osusky (PhD student from Slovak University of Technology, Bratislava, 17.04.08-17.05.08.)

Profile:

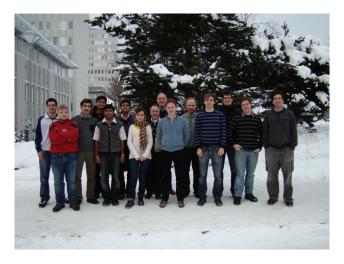
Process systems engineering deals with the overall system behaviour, and how the individual units should be combined to achieve optimal overall performance. Important topics are multi-scale process modelling, operation and control, design and synthesis, and simulation, statistics and optimization. The group closely cooperates with other systems-oriented departments at the university, including Engineering Cybernetics, Energy and Process Engineering, and Industrial Ecology, and also with SINTEF. The process systems engineering activity at NTNU (PROST) holds high international standards and was already in 1994 recognized as a strong-point centre, both by NTNU and SINTEF.

At present, the main activities in the group are within process control and process modelling including efficient thermodynamic calculations. A new interdisciplinary branch focusing on systems biology is currently being initiated enriching the NTNU biology effort with a systems component.



Control Activities:

Industrial use of advanced process control increases rapidly, and candidates who combine process knowledge and control expertise are in high demand in industry. Control is an enabling technology, thus basic for any industry-based society. The use of advanced control is transforming industries previously regarded as "lowtech" into "high-tech". In process control (Sigurd Skogestad and Heinz Preisig), the objective of the research is to develop simple yet rigorous tools to solve problems significant to industrial applications.



From left: Laszlo, Magnus, Mehdi, Deeptanshu, Ramprasad, Shams, Maryam, Heinz, Sigurd, Johannes, Tore, Jørgen, Ivan, Håkon, Nadi

Up to now, the design of the overall "plant-wide" control structure has been based on engineering experience and intuition, whilst the aim has been to develop rigorous techniques. The concept of "self-optimizing control" provides a basis for linking economic optimization and control (Sigurd Skogestad). For example, for a marathon runner, the heart rate may be a good "selfoptimizing" variable that may be kept constant in spite of uncertainty. Control is done in a hierarchical construct. At the bottom of the hierarchy, the main issue is to "stabilize" the operation and follow the setpoints provided by the layer above. Further up in the hierarchy one finds optimising control co-ordinating the control of units and plants. A special case is sequential control. which is used to implement recipes in batch operations but also is the basics of handling start-up and shut-down as well as all fault and emergency handling. Another important concept is controllability, which links control and design. Here the main focus is on applications, which currently include reactor and recycle processes, distillation columns, gas processing plants, cooling cycles including liquefied natural gas (LNG) plants, lowtemperature polymer fuel cells and anti-slug control. Small-scale experimental rigs have been built to study anti-slug control and novel distillation arrangements. In most cases, control is an "add-on" to enable and improve

operation, but the anti-slug rig demonstrates how control in some cases can be used to operate the system in a completely different manner.

Modelling Behaviours:

The centre piece of process systems engineering is the model. Modelling is seen as a difficult and time consuming operation. The step-wise approach developed in this group has transformed the art of modelling into a nearly procedural operation, which has been captured in a program environment. The modelling operation is thereby lifted up from writing equations to choosing concepts and mechanisms. The equations are then generated and assembled automatically taking the applicable equations from a data base that has built applying mechanistic descriptions where ever applicable. Multi-scale modelling is supported by enabling order-ofmagnitude assumptions, which automatically induce model reduction thereby eliminating structure-related mathematical problems. The overall objective in the group is to develop efficient object-oriented software tools that implement this method and assist in developing consistent and structurally solvable process models on different scales that match the particular application. The technology is physics-based with extensions to allow for grey-box modelling. It aims at replacing various graphical interfaces to simulators and generates code for the major chemical engineering simulators such as gProms, Matlab, Modelica etc. but will also be able to generate stand alone, applicationtailored simulators.

The fourth generation of a high-level modelling tool is presently being developed (Heinz Preisig), which we aim to apply to large-scale plants, including the Mongstad refinery. It incorporates object-oriented tools for efficient thermodynamic modelling, which extend into the efficient computation of thermodynamic information. Rather than a traditional implementation of activity or fugacity coefficients, emphasis is put on the use of structured equation sets governed by thermodynamic consistency rules (Haug-Warberg).

The thermodynamic models are implemented in symbolic form with automatic differentiation capabilities and serves as the basis of several industrial strength simulations (YASIM, CADAS) and energy accounting tools (HERE) in co-operation with StatoilHydro and Yara. A primary aspect of thermodynamic (and other physics) modelling is the required consistency of physical units. We have a procedure to obtain selfconsistent models, including automatic generation of gradients. This technique has so far been tested up to sixth order gradients, which are needed for higher-order critical point calculations.

In cooperation with Yara AS, Tore Haug-Warberg implemented a thermodynamic stream calculator "Yasim". It has a gentle learning curve using the familiar Excel worksheet interface whilst using state-of-the-art thermodynamic methods. All model information including mass balances, energy balances, chemical and phase equilibrium relations are defined in symbolic form. Differentiations are done in symbolic form. These properties add unsurpassed flexibility to Yasim that is not found in any other software of its kind. The ease of use should make it ideally suited for training and use in an industrial environment.

Model-Process Interface:

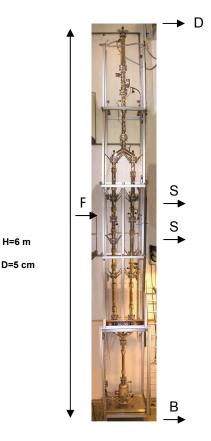
The model generally needs to be fitted to experimental data, and the group has always has a strong focus on statistical methods and experimental design (Hertzberg). Although Terje retired as professor in 2007, he is still active in this area, and in particular, in teaching.

Equipment:

The Kaibel distillation column (see picture) is 6 meter high and 5 cm in diameter and can be used to study "thermally coupled" columns, including the threeproduct Petlyuk column and the four-product Kaibel column. Dr. Ivar Halvorsen from SINTEF and Sigurd Skogestad manage this integrated distillation project. The group also has an automatic drink mixer, which is used for demonstration purposes and to study sequence control based on automata theory (Heinz Preisig).

Funding Resources:

Funding comes from the Norwegian Research Council, the Gas Technology Centre at NTNU and SINTEF, from industry (StatoilHydro, Gassco) and from the EU (ProMatch program).



Kaibel Distillation column.

ENVIRONMENTAL ENGINEERING AND REACTOR TECHNOLOGY GROUP

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Taek-Joong Kim Tom-Nils Nilsen

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Guests

Feng Qin (PhD student from Tsinghua University, China) Sinead Mc Dermott (PhD fra National University of Ireland, Maynooth, Ireland)

The two former groups in Separation and Environmental Technology and in Reactor Technology joined forces in 2008. The merger was motivated by similar research and teaching activities with respect to environmental engineering issues, especially in gas separation and capture of carbon dioxide. Joint projects and relocation in chemistry building 4 will serve to consolidate this cooperation in new laboratory facilities. Environmental engineering and reactor technology is the largest research group in the department covering interests in the fields of chemical reactor research, process design, acid gas absorption, membrane research, and crystallization and particle design.



Second row: Yuefa, Hallvard, Ameeya, Jens-Petter, Zhengjie, Marius, Eddie, Washim, Arne, Tom-Nils, Martin, Magne, Hugo Nils, Peter, Jon Arvid, Xiao, Ralf, Feng First row: Adrian, Qiang, Kumar, Kalim, Xuezhong, Inger Lise, Cathrine, Ellen, Ingvild, Andrea, Hanna, Inna, Arlinda, Erik, May-

Chemical Reactor Research

The activity on reactor engineering has been concentrated in fields directly supporting the design and development of chemical reactors and reactive separations. The most important research areas are:

Britt, Taek-Joong, Pablo, Arshad, Ugochukwu, Mehdi, Ahmad, Ali

- Mathematical modeling of chemical reactors.
- Multiphase flow modeling.
- Design of novel solution methods and algorithms.
- Experimental analyses of fluid flow, and heatand mass transfer phenomena in chemical reactors.
- Experimental validation of numerical models.
- Analysis and design of reactors for environmentally friendly chemical processes.

The research in these fields comprises both experimental and theoretical studies and we have a large range of well instrumented cold flow multi-phase reactors, as well as in-house software for multi-phase reactor simulations.

We are active users of Matlab, however, the computationally demanding models are implemented in FORTRAN 90 and C++. Application areas are special chemicals reactors, polymer production, sorption enhanced steam methane reforming, conventional synthesis gas and methanol synthesis, membrane reactors, wood gasification and chemical looping combustion.



A stirred tank used for studies of heat transfer and flow phenomena.

Educationally the main objective of our group is to educate MSc for the Norwegian industry and to raise the national scientific competence in our field of research through PhD studies.

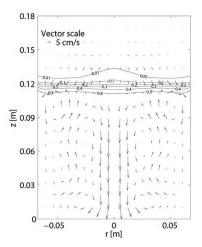
Research activities

The most important research projects are described in the following paragraphs. For a more comprehensive description, see our home pages: (http://www.chemeng.ntnu.no/research/reactmod/).

Modeling of multi-phase reactors

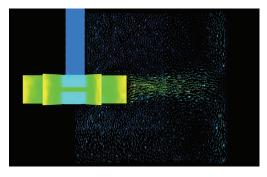
We have for more than 20 years been developing inhouse CFD codes for simulating multiphase flows in chemical reactors. Lately, our main focus has been put on developing modules for bubble/droplet break-up and coalescence within the population balance equation (PBE) framework.

The PBEs are solved accurately by efficient spectral and spectral-element methods designed for this particular purpose.



Flow pattern in a bubbling fluidized bed.

We are also investigating the performance of chemical reactive systems like fluidized beds, fixed bed reactors and agitated tanks. At present we are working with the design of suitable reactors for sorption enhanced reaction processes (SERP) like steam reforming with absorbents for CO_2 .



Simulation of the flow pattern from a turbine impeller.

The conventional 1D and 2D steady-state reactor models are normally run on standard PCs whereas the more computationally demanding dynamic 2D and 3D single and multiphase flow simulations are run on the national super-computers located at the university.

Advanced Modeling and Simulation of Chemical Reactors

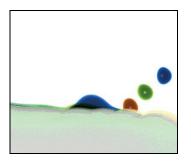
The activity in this area is mainly founded by the Norwegian Research Council through the GassMaks program. The work in this project is focused on modeling of chemical reactors like stirred tanks, fixed packed bed-, bubble column-, and fluidized bed reactors by the complete multifluid model containing a population balance equation for the fluid particle size distribution for the multiphase reactors. The model equations are solved by the modern least squares spectral element method. In the next phase of the project the novel in-house codes will be applied analyzing the chemical reactor processes utilizing natural gas as feedstock.

CO2 capture and droplet removal

Removal of droplets from high pressure gases An area of great importance for the Norwegian gas producing industry is the separating out of droplets from high pressure gases. Downstream process equipment e.g. compressors, separation processes or chemical reactors suffer disturbed operation or break-down if gases are not droplet free. Today's units are not good enough at high pressures, and robust and reliable solutions must be found in particular for sub-sea completions. The projects HiPGaS and now HiPGLS were established for studying the phenomena governing such separations. In collaboration with ten industrial partners separation rigs have been established for both low and high (<150 bar) pressures at NTNU and at the StatoilHydro research center. Tests within the project have also been performed at semi-industrial scale at the K-lab facility at Kårstø. Results from the high pressure rigs are unique and form a basis for model validation. In addition a laser laboratory is built for the study of droplet/droplet and droplet/surface collisions and for studies on the stability and break-up of liquid surfaces leading to re-entrainment of droplets. The studies are performed with a strong interaction between experiments and numerical models which run in either desktop computers or the national High Performance Computing resources.

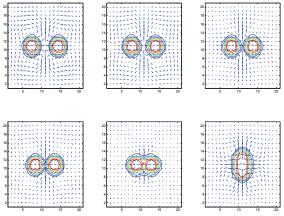
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Droplet collision leading to break-up. Experimental result to the left and model right.



Droplet generation from liquid film re-entrainment. Fake colors to identify different time steps.

 CO_2 capture from exhaust gases and natural gas. One of the most fundamental problems facing the earth today is global warming. The emissions of CO_2 , the most important greenhouse gas must be reduced, e.g. by CO_2 Capture and Storage (CCS). The research in this



Droplet collision leading to coalescence.

area comprises many projects, funded by the Research Council of Norway, the industry, and the European Union. Our work is concentrated along two axes, one studying CO₂ capture from off gases from fossil fueled power plants and from the iron and steel-making industry, and the other directed toward the removal of acid gases from natural gas. We have been heavily involved the EU FP6 projects, e.g the Integrated Project CASTOR, being in charge of developing new solvent systems for CO₂ capture from exhaust gases from coal and natural gas fired power stations. This work continues in the EU FP7 CESAR project. This work involves all the steps from theoretical screening by use of computational chemistry, through experimental screening, characterization of equilibria, thermal properties, transport properties and kinetics, to testing in

Process Design

A process design is to a large extent a consequence of developments on catalyst, choice of reaction routes, selection of solvent system, fluid type etc. At this level of development the structure of the chemical system and the kinetics are determined. Much research is focused on these topics because even incremental improvements may have large economical consequences. The next major step is to find a suitable reactor and process in which to deploy the system on a larger scale. The traditional way of doing design of new processes is by selecting reactor type and process configuration based on comparison to a similar known system. Design choices are often made on the basis of past experience or trial-and-error using laboratory tests and repeated simulations. These activities are necessary. However, it is not likely that the traditional way alone will lead to the best possible process configuration and design. Complementary tools and methods are needed to lead the design engineer onto the path of optimal design. Deviations from the optimal design will lead to unnecessary loss of product yield, unnecessary large volumes and loss of energy. Within the conceptual process design activity some of the most important choices are made, which have large consequences on the profitability and environmental loads of the final process

a laboratory pilot plant. In parallel we develop rigorous thermodynamic models and improved models for combined mass and heat transfer. In addition we have developed a full rate based simulator for the whole absorption/desorption process, CO2SIM. In the EU FP6 project CAPRICE, finished beginning of 2009, we are in charge of pilot plant data collection and simulation software validation. The largest project on CO₂ capture is SOLVit, a JIP with Aker Clean Carbon (ACC). Other projects in this area are EU FP6 ULCOS, and the Norwegian Research Council financed BIGCO2, now BIGCCS. In the CCERT project, a JIP with four industrial partners, fundamental problems are addressed.



Laboratory pilot plant for CO₂ capture by absorption

technology. According to Douglas $(1988)^1$ the conceptual design of an integrated plant can be broken down into a hierachy of decisions and organized into different levels of activities. Among the levels of activities are reactor-separator-recycle structures, heat integration, and separation train sequence design. A method in focus here is a systematic procedure based on shortcut models. A path is a line of production on which basic operations or functions take place. Reactants pass through a series of functions or basic operations to form the desired products. The basic operations are represented by design functions on the volume path. The design functions are fluid mixing (dispersion), distribution of extra feed points, distribution of heat transfer area and coolant temperature, catalyst dilution distribution and more. The conceptual reactor design problem is solved as an optimal control problem. Parameterization of the design functions and the state variables are applied. The realization is a staged process string of multifunctional units.

Douglas, J. M., Conceptual Design of Chemical Processes, McGraw Hill, 1988.

Membrane Research

Membranes for gas separation and osmotic processes

(more information about the Memfo (Membrane Research activities): www.chemeng.ntnu.no/memfo

The membrane research programs have extensive activities both on basic membrane material development, as well as membrane gas separation processes, modelling and simulations. The main focus for the research is CO_2 capture by membranes (from flue gas, natural gas sweetening, biogas upgrading) and hydrogen recovery from various mixed gas streams. In addition to these energy focused gas applications, there is also ongoing research on membranes for chlorine separation. The membrane materials in focus are various types of polymers, nano-composites, carbon membranes, and modified glass membranes. The international network is extensive, with cooperation both within EU-projects, USA, Japan, the Nordic countries and Russia.

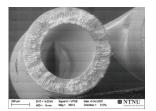
Brief description of sample gas separation projects

a) EU FP6 project NaturalHy (www.naturalhy.net) The project involves about 40 partners, and the main idea is to investigate the possibility of using the European gas network for co-transport of hydrogen with natural gas. Realising that the society only slowly will convert to hydrogen based energy, the idea is that hydrogen may be injected into the gas net at various places where it is produced, and then separated from the natural gas at points along the line where pure hydrogen is needed (e.g. for fuel cells). The task into which Memfo is doing research, is development of a suitable membrane for hydrogen recovery. Carbon molecular sieve membranes are investigated for this purpose. One post doc is engaged in the research, and very promising results have been documented. The project continues for 1 more year. The project coordinator is Gasunie in the Netherlands.

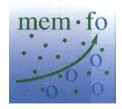
b) KMB RENERGI / NFR project CEPEME Membrane development for selective CO₂ capture

The membrane material being developed in this project contains a specific "carrier" which makes it selective for CO₂ while other gas components are being retained. The material is based on a polymer containing fixed amine groups as carriers. A significant progress has been achieved during the last couple of years, both with respect to CO₂-flux and selectivity compared to the other components in a mixed gas. The obtained results have drawn international attention. The c) EU FP6 project NanoGloWa (www.nanoglowa.com) The project started November 2006 and is focusing on "Nano-structured membranes against Global Warming" (NanoGloWa). Memfo is a major partner in the project; task leader for two work packages focusing on development of carbon membranes and polymeric materials for CO₂ capture from power plants. The

membrane is patented, and there are big expectations for the further development; for CO2 capture from coal fired or gas fired power plants as well as other CO2 containing gas streams. The project ended in 2008. Preparation for small scale pilot scale testing is now being done. One research scientist and one PhD student have been working on the project. A focus on process development has also been initiated. Project partners further are NFR and StatoilHydro; negotiations with membrane producers are ongoing for the scale up. There is also interest from industry on other applications where CO_2 is present in the gas stream (natural gas sweetening, CO_2 removal from anaesthetic gas...)



<u>SEM-picture</u> showing a membrane. DO-DI: 1 -0.6 mm cut through a hollow fibre



Memfo = Membrane Forskning (Membrane Research)



Bundle of hollow fibre membranes in an industrial module-packing densities can be up to 30 000 m^2/m^3

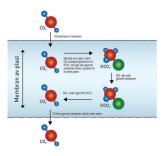


Illustration of the mechanism in the patented CO₂ facilitated transport membrane

project also includes spinning of hollow fibres, module development and durability tests. One Post.doc. and one PhD student were attached to the project in 2008. The project coordinbator is KEMA in The Netherlands.

d) EU FP7 project DECARBit

The project is coordinated by Sintef and had kick-off in Feb 2008. The focus is pre-combustion separation of CO_2 and H_2 at high temperatures (>300°C). Carbon membranes are being tailored for this separation. The project also includes module development and durability tests. The project coordinator is SINTEF Energy

e) KMB GASSMAKS / NFR project RECCO2

The project started late 2007. The objective is CO_2 removal from high pressure natural gas streams using a polymeric blend membrane; using as one of the polymers in the blend the patented PVAm facilitated transport membrane. The project includes material development, pilot construction, durability tests and simulations. The main challenge in this project is the performance at high pressures (\rightarrow 100 bar). Special restrictions for HMS is needed in this operating range. Industrial partner is StatoilHydro.

f) KMB NANOMAT / NFR project development of nanocomposite membranes

These projects are in cooperation with Sintef and North Carolina State University. The material development for hydrogen – CO_2 separation are in focus. There were one PhD-student on the US-side; one PhD and one Post doc on the NTNU-side. The materials studied at NTNU were nano-composites (mixed matrix of high free volume polymers and nanoparticles) and block copolymers on the US-side – very challenging research work. Within this project a NASA award was granted during 2005 for a US patent. The project ended in 2008.

g) FP / NFR Development of hybrid membrane for chlorine purification

Memfo is one of very few groups in the world doing research on membranes for the purification of chlorine gas. This is probably for security and safety reasons, and the challenge of handling this poisonous gas. The project is extremely challenging, but if successful, it will be a major step towards simplification of expensive and complicated unit operations for recovery of chlorine from various process streams. Chlorine is one of the major chemicals used in chemical process industry worldwide. The materials in focus are glass and perfluorinated polymers. There is co-operation with Japanese research in this project. Small steps forward have been documented during 2008, and one post doc finished his work on this project in 2008. There is a major interest from industry in this project.

<u>Brief description of a sample liquid separation project</u> KMB NANOMAT / NFR project: Membranes for osmotic processes – POPMOP

Reverse osmosis is today one of the major methods for desalting ocean water into freshwater. Osmotic effects demands that a high pressure on the saltwater side of the membrane modules must be applied to override the osmotic pressure difference between sea water and freshwater. If the applied pressure in the cells is lower than the osmotic pressure the water flow is reversed and freshwater flows into the saltwater compartment, thus increasing the volume of moderately pressurized saltwater. This is the principle of Pressure Retarded Osmosis (PRO), which may be used to produce electric energy as the surplus water on the saltwater side may be run through turbines for power production. The potential for power production at the outlet of every river that flows into the ocean is very large. Towards the end of 2007 Memfo was granted a KMB project from Nanomat/NFR supported by Aqualyng, Statkraft and StatoilHydro. The project is aiming at

studying and optimizing the synthesis of polymers for osmotic processes. This development includes both ptimisation of the support membrane as well as the thin film membrane itself. Two PhD students and one research scientist have been working on the project in 2008

<u>Various</u>

The Memfo researchers work very much as a team in addition to be dedicated to individual projects. Hence the simulation of processes is handled whenever needed by those who have the competence. Likewise; the concern for environmental issues, leads to the focus also on biogas upgrading to vehicle fuel quality (biomethane). Carbon membranes has proved to be suitable for upgrading of biogas; documented by experiments and discussed in publications.

A collaboration with University of Dar es Salaam (UDSM), Tanzania, has resulted in a project funded by NUFU. The focus is on bioenergy; with certain aspects of the production of bioethanol and biodiesel from local raw materials. There are 2 PhDs at UDSM and 1 at NTNU attached to this project.

There are several well equipped laboratories available for the membrane research; both for membrane material development as well as membrane characterization. The membrane research on CCS is especially strong, and completes the research on CCS activities in collaboration within the environment and reactor group.

A spin-off company from the research on carbon molecular sieve membranes was established in 2008 with the support of the Technology Transfer Office (TTO) at NTNU. The name of the company is <u>MemfoACT</u> (= Membranes for Advanced Clean Technology, <u>http://www.memfoact.no</u>)

Crystallizaton and Particle Design

The research within crystallization is focused at kinetics of nucleation, crystal growth, and agglomeration in order to predict and control the particle size distribution and shape of crystalline particulate products for scale prevention, improved solid-liquid separation, CO₂capture in precipitating systems, and for nano-particle production. The crystallization group also investigates fundamental mechanisms in the early formation of solid particles as well as mechanisms for growth of polycrystalline particles. **Optimisation of Glycol Loop Design and Operation** The aim of the project is to develop a simulation tool for glycol loops in processing of natural gas. This necessitates a deep understanding of the precipitation and crystallisation behaviour of salts and scale-forming carbonates in ethylene glycol (MEG) and water mixtures. Kinetics of calcium carbonate precipitation in the glycol injection point off-shore and the crystallization and separation of salts in the on-shore glycol reclamation units will be the main research tasks. The project is in collaboration with Institute of Energy Technology, Norway (IFE) and financed by several international oil and gas companies and the Research Council of Norway (NFR). Glycol injection point off-shore and the crystallization and separation of salts in the on-shore glycol reclamation units will be the main research tasks. Studies performed in the crystallization group in 2008 have shown that the MEG significantly affects the precipitation of calcium carbonate by lowering the growth rate, promoting nucleation and by shifting the polymorphic composition. Kinetic expressions have been developed to be implemented into computer simulator for particle formation control within glycol loops.

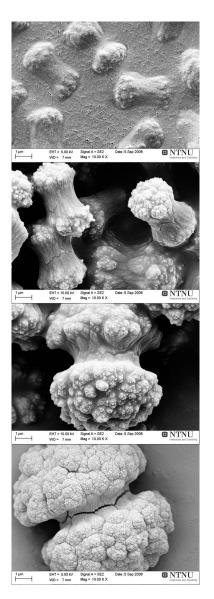
Industrial Crystallization and Powder Technology

The goal of this project is to relate solid/liquid separation characteristics and resulting dry powder flow properties to the underlying growth and agglomeration phenomena. It involves studies of inorganic salts and pharmaceuticals, and the primary target for the activity at NTNU is to link the parameters in the crystallisation process to the subsequent filtration step by focusing on common mechanisms for these selected systems. The project is in collaboration with POSTEC at Tel-Tek and is financed by the Research Council of Norway (NFR) and Norwegian industry partners.

In 2008 we have investigated the effect of supersaturation and temperature on the particle design of pharmaceutical compounds. We have identified a general mechanism of crystal growth switching whereby the particle shape is dramatically altered. This is illustrated for the precipitation of sodium glutamate by switching from the well-known needle crystals of β -glutamic acid to spherical particles of the same polymorph. Filtration resistance measurements as well as powder flow properties measured by uniaxial testing has shown that these spherical particles are unwanted, and in some cases the conditions can be met to avoid their formation.

Investigations of spherulitic growth in solutions – a mechanism for polycrystalline particles

During 2008 we have collected experimental proof for spherulitic growth from solutions in contrast to literature claims of nano-aggregation as responsible for the same type of particles. Spherulitic growth is usually associated with melt crystallization but our findings show that it is a general mechanism is solid material formation. The time-dependent evolution of spherulitic growth is shown for calcium carbonate in the figure below (the scale bar equals one micrometer) and work is in progress to prove the validity of this growth process for numerous crystalline compounds, irrespective of the chemical nature of the consituents.



The evolution of spherulitic growth of polycrystalline particles of calcium carbonate 15, 25, 45, and 140 minutes after nucleation.

PAPER AND FIBRE TECHNOLOGY GROUP

Academic staff

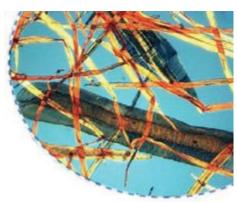
Professor Øyvind W. Gregersen Associate professor Størker Moe Professor emeritus Torbjørn Helle

Post docs

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PhD candidates

David Vaaler (until 30.05.08) Galina Rodionova (from 01.12.08) Håkon Nordhagen Marianne Lenes Marius Rusu Mihaela Tanase (from 01.01.08) Sara Paunonen



The size distributions of fibres and fines are essential for the papermaking properties of a pulp.

Teaching

The Paper and Fiber Technology group provides chemical engineers and PhDs for the Norwegian pulp and paper industry. The estimated need from the industry is 8-10 engineering graduates and about 2 PhD candidates per year. During 2006 3 MSc and 4 PhD candidates graduated from our group. We also gave two industry courses in Pulp and Paper technology and a course in pulp and paper technology at The Norwegian University of Life Sciences (UMB).

Partners

The Paper and Fibre group, Paper and Fibre research Institute (PFI) and parts of the Ugelstad laboratory (colloid and surface chemistry) are located in the same building on the NTNU

Gløshaugen campus and are working in close cooperation. We also cooperate closely with pulp and paper industry partners such as Norske Skog, Södra Cell, Borregaard, Peterson and Voith.

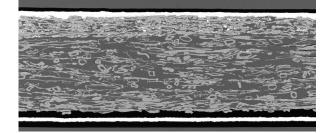
Research

Our research is focused on improvements in the pulp and paper process and on improved end product quality. Examples are:

- Improved runnability of printing paper by use of fracture mechanics as a tool in paper quality optimization.
- Reduced energy consumption and/or better fibre properties of mechanical pulp through high intensity refining, co-refining of different raw materials or pre-treatment of wood chips.
- Reduced print through defect in newsprint by optimized sheet structure
- Better strength and surface properties of woodcontaining paper by use of micro-fibrillar cellulose as an additive.
- The influence of paper structure on mechanical, optical and surface properties of paper.

Trends

The last years a new activity on use of cellulose based particles in composite materials have started. Both better oxygen barrier and mechanical properties may be obtained by using cellulose fibres or fibrils as reinforcement in thermoplastic composites. Together with PFI a new research activity on the use of wood based bio-fuel production have also started. The goal is cost effective production of bio-diesel and ethanol from wood. Further, energy effective production is one of the primary concerns of the pulp and paper industry and thus also an important research area now and in the future.



The mechanical and barrier properties of a paper material depend on the material choice and structure. The SEM image of the cross section of a liquid board shows (from top) a Polyethylene (PE), mineral coating, kraft pulp, CTMP, kraft pulp, PE, Aluminium layer and PE.

CHAPTER 3: PUBLICATIONS

PUBLICATIONS IN REFEREED JOURNALS

1. Amirkhani, Masoud; Volden, Sondre; Zhu, Kaizheng; Glomm, Wilhelm; Nyström, Bo. Adsorption of cellulose derivatives on flat gold surfaces and on spherical gold particles. *Journal of Colloid and Interface Science* 2008 ;Volum 328.

2. Antoine, Christine; Gregersen, Øyvind Weiby. On the three component quantification of print-throud. Part 2: Effect of the ink vehicle extraction on print optical properties. *Nordic Pulp & Paper Research Journal* 2008 ;Volum 23.(1) s. 29-33

<u>3</u>. Araujo, Antonio Carlos Brandao; Skogestad, Sigurd. Control structure design for the ammonia synthesis process. *Computers and Chemical Engineering* 2008 ;Volum 32.(12) s. 2920-2932

4. Aske, Elvira Marie B; Strand, Stig; Skogestad, Sigurd. Coordinator MPC for maximizing plant throughput. *Computers and Chemical Engineering* 2008 ;Volum 32. s. 195-204

5. Aske, Elvira Marie B; Strand, Stig; Skogestad, Sigurd. Implementation of coordinator MPC on a large-scale gas plant. *Elsevier IFAC Publications / IFAC Proceedings series* 2008 s. -

6. Baldea, M; Araujo, Antonio Carlos Brandao; Skogestad, Sigurd; Daoutidis, P. Dynamic considerations in the synthesis of self-optimizing control structures. *AIChE Journal* 2008; Volum 54. s. 1830-1841

7. Bar, Nadav; Lale, Rahmi.

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140. Zhu, Zhengjie; Patruno, Luciano E; Dorao, Carlos Alberto; Lucas, Dirk; Jakobsen, Hugo Atle. Simulation of bubble coalescence in bubble column using the least-squares method. 11th International Conference on Multiphase Flow in Industrial Plants; 2008-09-07 - 2008-09-10

CHAPTER 4: EDUCATION

Chemical Engineering

The specialization in Chemical Engineering starts in the third year with the basic technological courses in Separation Technology, Reaction Engineering, Thermodynamics and Process Design. In the fourth year the students elect further specializations for the remaining of the studies. The students choose between 5 specializations: Petrochemistry and Catalysis, Colloid and Polymer Chemistry, Environmental Engineering and Reactor Technology, Process Systems Engineering and Paper and Fibre Technology. The goal of the education is a Master (MSc) at a high international level in Chemical Engineering.

Students with a bachelor degree in a relevant area from colleges can be admitted to the fourth year of the MScdegree programme (2 year MSc programme). The degree provides the candidates qualifications for jobs in a wide range of industries, as well as the public sector and in research. It is also the basis for admission to the PhD-studies in Chemical Engineering.

Master courses given in 2008:

Course code				Registered	Passed
TKP4100	Fluid Flow and Heat Transfer	7.5		79	65
TKP4105	Separation Technology	7.5		56	44
TKP4110	Chemical Reaction Engineering	7.5		74	55
TKP4115	Surface and Colloid Chemistry	7.5		79	68
TKP4120	Process Engineering	7.5	2	129	113
TKP4125	Paper and Fiber Technology	7.5	4	5	4
TKP4130	Polymer Chemistry	7.5	4	17	14
TKP4135	Chemical Process Dynamics and Optimization	7.5	4	6	4
TKP4140	Process Control	7.5	4	35	31
TKP4145	Reactor Technology	7.5		10	9
TKP4150	Petrochemistry and Oil Refining	7.5		24	22
TKP4155	Reaction Kinetics and Catalysis	7.5		47	37
TKP4160	Transport Phenomena	7.5		30	23
TKP4165	Process Design	7.5		25	22
TKP4170	Process Design, Project (autumn)	7.5		24	24
TKP4171	Process Design, Project (spring)	7.5		8	8
TKP4175	Thermodynamic Methods	7.5	3	36	25
TKP4510	Catalysis and Petrochemistry, Specialization Project	15	5	7	7
TKP4511	Catalysis and Petrochemistry, Specialization Project	7.5	5	0	0
TKP4515	Catalysis and Petrochemistry, Specialization Course	7.5	5	10	10
TKP4520	Colloid and Polymer Chemistry, Specialization	15	5	3	3
	Project				
TKP4521	Colloid and Polymer Chemistry, Specialization	7,5	5	1	1
	Project				
TKP4515	Colloid and Polymer Chemistry, Specialization	7.5	5	4	4
	Course				
TKP4530	Reactor Technology, Specialization Project	15	5	3	3
TKP4531	Reactor Technology, Specialization Project	7,5		0	0
TKP4535	Reactor Technology, Specialization Course	7,5		3	3
TKP4540	Separation and Environmental Technology,	15	5	9	9
	Specialization Project				
TKP4541	Separation and Environmental Technology,	7.5	5	1	1
	Specialization Project				
TKP4545	Separation and Environmental Technology,	7.5	5	14	14
	Specialization Course				
TKP4550	Process Systems Engineering, Specialization	15	5	3	3
	Project				
TKP4551	Process Systems Engineering, Specialization Project			0	0
TKP4555	Process Systems Engineering, Specialization Course		5	3	3
TKP4560	Paper and Fibertechnology, Specialization Project	15	5	1	1
TKP4561	Paper and Fibertechnology, Specialization Project	7.5		0	0
TKP4565	Paper and Fibertechnology, Specialization Course	7.5	5	1	1
TKP4850	Experts in Team, Interdisciplinary Project	7.5	4	18	18
TKP4500	Final Year Design, for Spanish students	60	5	3	3
TKP4900	Chemical Process Technology, Master Thesis	30	5	31	31

Master theses 2008 Total 32, 18 female and 14 male

Beinset, Morten Sorption Enhanced Steam Glycerol Reforming Supervisor: De Chen

Bekkevold, Jan Petter Hydrogen sorption enhanced water-gas shift using metals forming hydrides as sorbents Supervisor: Hilde Venvik

Braathen, Bjarne Wax deposition in crude oil systems Supervisor: Johan Sjöblom

Elde, Ingrid Elise Model studies of inhibition of hydrotreating catalysts Supervisor: Edd A. Blekkan

Ellingsen, Christian

Safety constraints in an artifical β-cell: an implementation of Model Predictive Control (MPC) with Insulin-on-Board (IOB) Supervisor: Sigurd Skogestad

Evensen, Trond The use of gas hydrates for water purification Supervisor: Jens-Petter Andreassen

Fagerbekk, Siri Albertsen New solid acceptors for CO₂ Capture Supervisor: De Chen

Fahadi, Jalal

Dynamic simulation for improved operation of the Snøhvit CO₂-removal section Supervisor: Sigurd Skogestad

Fostenes, Siv Monica

The Effect of Scaling Inhibitors on the Precipitation of Calcium Carbonate in Solutions of Ethylene Glykol and Water. Supervisor: Jens-Petter Andreassen

Frøseth, Fredrik

Partial oxidation of methane over perovskites Supervisor: Anders Holmen

Fævelen, Erlend Schou

Kinetics of Nucleation and Crystal Growth of Calcium Carbonate in Mixtures of Mono Ethylene Glycol and Water Supervisor: Jens-Petter Andreassen

Haukebø, Siv Hustad

Production and characerization of selective hollow fiber membranes for CO_2 capture Supervisor: May-Britt Hägg

Husås, Ranveig

Characterization of asphaltenes and maltenes in feed and product oil samles from a pilot plant for heavy oil upgrading Supervisor: Edd A. Blekkan Jøndahl, Mari

Microchannel membrane reactor for production of hydrogen by methanol steam reforming Supervisor: Hilde J. Venvik

Kleppa, Gøril

Synthesis and Characterization of Nano Structured Alumina Materials Supervisor: Gisle Øye

Knudsen, Agnethe

Synthesis of nanostructurated monoliths for the Fischer-Tropsch catalyst Supervisor: Gisle Øye

Kornberg, Anette

Dynamic Simulation and Control of Ethane/Carbon Dioxide Plant Supervisor: Sigurd Skogestad

Lavalle Jambert, Maria

Testing and analysis of suitable membranes for osmotic power Supervisor: May-Britt Hägg

Lie, Marianne

Advanced characterization of particles in produced water Supervisor: Gisle Øye

Machado Miguens, Carolina Andrea

Methanol Synthesis in Microchannel Reactor Supervisor: Hilde Venvik

Opedal, Nils van der Tuuk

Synthesis of metallic nano structures Supervisor: Gisle Øye

Pettersen ,Tone Sejnæs

Synamic mode of offshore processing plant with emphasis on estimating environmental impact Supervisor: Sigurd Skogestad

Salamanca Parra, Jacobo Manuel

Studies of hydrogen production by catalytic steam reforming of biomass-related compounds Supervisor: Edd A. Blekkan

Skog, Tom-Gøran

Durability of membrane material when exposed to impurities and higher hydrocarbons in natural gas Supervisor: May-Britt Hägg

Skogestad, Hanne

Engineering membrane selectivity for CO₂ separation Supervisor: May-Britt Hägg

Smedsrud, Helge

Dynamic modelling and Control of Brobekk Incineration Plant Supervisor: Sigurd Skogestad *Spets, Øyvind* Crude oil properties: Relation between viscosity and conductivity Supervisor: Johan Sjöblom

Tveten, Erik Zakarias Studies of alumina based cobalt Fischer-Tropsch catalysts Supervisor: Anders Holmen

Vatneberg, Stine Vemmestad Scaling of Calcium Carbonate on Metal Surfaces in Mixtures of Mono Ethylene Glycol and Water Supervisor: Jens-Petter Andreassen

Uwarwema, Theogene Use of dynamic degrees of freedom for thighter bottleneck control Supervisor: Sigurd Skogestad

Zhu, Ye Synthesis of nano crystalline oxides – carbon nanofibers composite Supervisor: De Chen

Aarhoug, Kristin Characterization of oil in produced water Supervisor: Gisle Øye

5rd year students 2008/2009 Total 32, 22 female and 10 male

Aulie, Martin Håkon Bergstedt, Elin Biørn. Inger Lise Bjartnes, Kirsti Enaasen, Nina Gran, Håvard Foss Heskestad, Rita Holsæter, Hege Christine Høyen, Ragnhild Haaversen, Linn Christine Loe Johansen, Hege Døvle Jonassen, Øystein Kalstad, Tone Catherine Hval Karlsen Kolås, Kjersti Blytt-Tøsdal Lin, Yi Løvold, Ane Dyb Munkebye, Knut-Arne Rademacher Mæhle, Inger Roksvåg Nenningsland, Andreas Lyng Røsting, Kristine Sjulstad, Johanne Schjellungen Solsvik, Jannike Storsæter, Kathrine Trehjørningen, Kristin Rem Tuvnes, Henrik J. Tørneng, Eirik Fatnes Vaktdal, Hanne Margrethe Vattekar, Petter Tangen Vik, Camilla Berge Zhang, Xi Aaserud, Jo

4rd year students 2008/2009

Total 22, 10 female and 12 male

Almeland, Silje Kreken Eklud, Pål Helberg, Ragne Marie Lilleby Helgestad, Dag-Erik Hesstvedt, Camilla Hyllestad, Elisabeth Lovise R. Jacobsen, Julie Berild Jens, Christian Morten Jensen, Martin Buus Lillebø, Andreas Helland Lunde, Magnus Munkejord, June Nergård, Liv-Turid Ringstad, Renate Roel, Carl Marius Roll, Sebastian Roset, Stine Thysnes Røed, Anders Haukvik Skarsgard, Torstein Gaarder Skjeldestad, Kjetil Tvedt, Arne Hetland Østbye, Helene

3rd year students 2008/2009

Total 22, 12 female and 10 male

Barland, Astrid Odland Berstad, Eivind Bøhn, Kristian Evenrud, Vegar Graff, Vidar Grimholt, Chriss Tony Robert Haglund, Ola Kjølberg Hareide, Henning Heggvoll, Børge Helgesen, Anette Hoel Håseth. Jenny Kristin Karlsen, Aina Elin Karlsen, Stine Kjos-Hanssen, Øyvind Jacob Krossholm, Charlotte Kvam, Torunn Leithe, Anna Elise Nergaard, Margrethe Osmani, Kimete Plünnecke, Katrine S. Biesterfeld Shadman, Amir Williams, Rebecca Sian

International master students 2008/2009

Total 9, 3 female and 6 male

Adedeji, Aderonke Nigeria Arthur, Theophilus Ghana Asif, Naveed Pakistan Gao, Bicheng Kina Gondal, Shahla Pakistan Haider, Shamim Pakistan Mba, Emmanuel Nigeria Osatiashtiani, Amin Iran Widarena, Trimaharika Indonesia

Student exchange 2008

26 exchange students visited our Department (12 females and 14 males)

Bombardier, Marion	Ecole Nationale Supéure (ENSIACET) France	React
Cabonargi, Ander Bengoechea	University of Basque Country, Spain	Separ
Cos, Francois	IUT A & Lille, France	Collo
Diou, Odile	Ecole Nationale Superieure de Chimie et	
	Physique, France	Collo
Equzkiagirre, Ana Cagligas	Universidad del Pais Vasco, Spain	Separ
Freeman, Stephanie	The University of Texas, Austin, USA	React
Garcia Borreguero,		
Jaime Hortelano	Universidad Autónoma de Madrid, Spain	Separ
Gaya, Eva Mir	Universitat Politecnica de Catalunya, Spain	Paper
Gutierrez, Bartolome Lopez	Universidad de Granada, Spain	Catal
Hacquard, Sebastien	Université Franche-Comté Besancon, France	Collo
Hebben, Nicole	University of Karlsruhe, Germany	Catal
Hoffmann, Martin	Technische Universität, Berlin, Germany	React
Ladner, Yoann	Université Franche-Comté Besancon, France	Collo
Lavall Jambert, Maria	Universitat Politechnica de Cataluynya, Spain	Separ
Leong, Chi Hoong	National University of Singapore, Singapore	Separ
Mc Dermott, Sinead	National University of Ireland, Maynooth	Separ
Miguens, Andrea C. M.	Universidad Simón Bolivar, Venezuela	Catal
Perez Aguilera, Luber C.	Simon Bolivar University, Venezuela	Proce
Repka, Martin	Brno University of Technology, Czech Republic	Separ
Salamanca Parra, Jacobo	University of Valladolid, Spain	Catal
Salas, Jérémie	Université Franche-Comté Besancon, France	Collo
Sauvage, Rémi	IUT A & Lille, France	Collo
Schneider, René	Otto-von-Guericke University, Germany	Proce
Vera Lozada, Rafael	Simon Bolivar University, Venezuela	Catal
Vuillemin, Marlene	INSA, France	
Zscherpe, Tina	University of Karlsruhe, Germany	Catal

Reactor Technology Group Separation and Environmental Group Colloid- and Polymer Group

Colloid- and Polymer Group Separation and Environmental Group Reactor Technology Group

aration and Environmental Group er and Fibre Technology Group lysis and Petrochemistry Group oid- and Polymer Group lysis and Petrochemistry Group ctor Technology Group oid- and Polymer Group aration and Environmental Group aration and Environmental Group aration and Environmental Group lysis and Petrochemistry Group ess System Engineering Group aration and Environmental Group lysis and Petrochemistry Group oid- and Polymer Group oid- and Polymer Group ess System Engineering Group lysis and Petrochemistry Group

Catalysis and Petrochemistry Group

13 students from our Department (9 females and 4 males) on exchange

Name

Haaversen, Linn Christine Loe Helgestad, Dag-Erik Hesstvedt, Camilla Holsæter, Hege Christine Jacobsen, Julie Berild Johansen, Hege Døvle Munkejord, June Skogestad, Hanne Tørneng, Erik Fatnes Tuvnes, Henrik J. Vaktdal, Hanne Margrethe Vattekar, Petter Tangen

To Institution

University of Granada, Spain			
University of Berkley, California, USA			
Curtin Univ. of Technology, Perth, Australia			
University of Bath, England			
University of California, Santa Barbara, USA			
Curtin Univ. of Technology, Perth, Australia			
University of California, Santa Barbara, USA			
University of Arizona, Tucson, USA			
University of California, Santa Barbara, USA			
University of California, Santa Barbara, USA			
Ecole Nationale Supéure (ENSIACET) France			
University of Calgary, Canada			

Programme	Period
Erasmus	15.08.07 - 30.06.08
Bilateral	21.08.08 - 21.05.09
Individ	01.07.08 - 01.07.09
Erasmus	15.08.07 - 30.06.08
Individ	22.09.08 - 30.06.09
Individ	01.01.08 - 30.06.08
Individ	01.09.08 - 30.06.09
Individ	04.02.08 - 20.07.08
Individ	01.09.07 - 30.06.08
Individ	15.08.07 - 30.06.08
Erasmus	15.08.08 - 30.06.08
Individ	15.08.07-30.06.08

PhD in Chemical Engineering

PhD courses given at Department of Chemical Engineering:

Cours code	Course title	Credits	Registered	Passed	Semester
KP8100	Advanced Process Simulation	7.5			
KP8102	Wood Chemistry in Pulping and Paper Making	7.5			
KP8104	Industrial Crystallization and Precipitation	9.0			Fall 08
KP8105	Methematical Modelling and Model Fitting	7.5	1	1	
KP8106	Gas Cleaning with Chemical Solvents	9.0	4	4	Spring 08
KP8107	Advanced Course in Membrane Separation Process	9.0			
KP8108	Advanced Thermodynamics: With applications to				
	Phase and Reaction Equilibria	9.0	1	1	Fall 08
KP8109	Environmental Catalysis	6.0	4	4	Spring 08
KP8110	Membrane Gas Purification	9.0	2	1	Spring 08
KP8112	Applied Heterogeneous Catalysis	6.0	3	2	Spring 08
KP8113	Characterizaton of Heterogeneous Catalysts	6.0	22	22	Fall 08
KP8115	Advanced Process Control	7.5	4	3	Fall 08
KP8117	Paper Physics and Paper Chemistry	9.0	2	1	Spring 08
KP8118	Advanced Reactor Modelling	12.0			
KP8119	Surfactants and Polymers in Aquous Soutions	9.0	9	7	Fall 08
KP8120	Colloid Chemistry and Functional Materials	9.0			
KP8121	Surface, Colloid and Polymer Chemistry,				
	Special Topics	6.0	9	7	Fall 08
KP8123	Colloid Chemistry for Process Industry	9.0	6	6	Spring 08
KP8124	Modelling of Catalytic Reactions	6.0	8	7	Fall 08
KP8125	Design and Preparation of Catalytic Materials	6.0			Fall 08
KP8901	Chemical Process System Engieering	7.5			Fall 08
KP8902	Reactor Technology	7.5			Fall 08
KP8903	Reaction Kinetics and Catalysis	7.5	3	2	Fall 08
KP8904	Transport Phenomena	7.5	1	1	Fall 08

PhD-Thesis 2008

Total 13, 6 female and 7 male

Dudasova, Dorota

Characterization of solid particle suspensions with organic coatings in oilfield produced water. Supervisor: Johan Sjöblom

Enger, Bjørn Christian

Hydrogen production by catalytic partial oxidation of methane Supervisor: Anders Holmen

Hammer, Nina

Au-TiO₂ catalysts supported on carbon nanostructures for CO removal reactions Supervisor: Magnus Rønning

Håkonsen, Silje Fosse

Oxidative dehydrogenation of ethane at short contact times. Supervisor: Anders Holmen

Jensen, Jørgen Bauck

Optimal Operation of Refrigeration Cycles. Supervisor: Sigurd Skogestad

Less, Simone

Mechanisms of water-in-crude oil emulsion formation, stabilization and resolution by electrostatic means. Supervisor: Johan Sjöblom

Lindborg, Håvard

Modeling and Simulations of Reactive Two-Phase Flows in Fluidized Beds. Supervisor: Hugo A. Jakobsen

Løvfall, Bjørn Tore

Computer Realization of Thermodynamic Models Using Algebraic Objects Supervisor: Tore Haug-Warberg

Meland, Hilde

Preparation and characterization of Cu- and Pt-based water-gas shift catalysts Supervisor: Anders Holmen

Shao, Lei

Crosslinking and Stabilization of High Free Volume Polymers for Gas Separations Supervisor: May-Britt Hägg Silset, Anne Emulsions (w/o and o/w) of Heavy Crude Oils. Characterization, Stabilization, Destabilization and Produced Water Quality Supervisor: Johan Sjöblom

Sivertsen, Heidi

Stabilization of desired flow regimes using active control Supervisor: Sigurd Skogestad Vaaler, David Yield-increasing additives in kraft pulping: Effect on carbohydrate retention, composition and handsheet properties. Supervisor: Størker Moe

7 PhD exchange students visited our Department in 2008 (1 female and 5 male).

Name	University/Country	Group located
Bimbela, Fernando	Universidad de Zaragoza, Spain	Catalysis Group
den Breejen, Johan	Utrecht University, Belgium	Catalysis Group
Lögdberg, Sara	KTH, Sweden	Catalysis Group
Ousky, Jakub	Slovak University of Technology, Bratislava	Process Systems Engineering Group
Qin, Feng	Tsinghua University, Beijing, China	Reactortechnology Group
Skudas, Romas	Johannes Gutenberg Univ., Mainz, Germany	Catalysis Group
Lualdi, Mateo	KTH, Sweden	Catalysis Group

1 PhD student from our Department visited other Universities in 2008

Name	University/Country	Period
Patanou, Eleni	IRCE, Lyon, France	18.11.08 - 09.12.08

Suplementery education

EVU-course KP6002 Paper Technology, 14.01. – 18.01, and 11. 02. – 15.02.2008. There were 17 participants, those who passed the exam got 6 ECTS. Responsible for the course: Professor Øyvind Gregersen

Seminars and meetings organized by the Department in 2008

Seminars:

KinCat Seminar in Heterogeneous Catalysis: "Biomass Fed Fischer-Tropsch System" Dr. Farshild Owrang, Chalmers, Gothenburg, Sweden, 28.03.2008

KinCat Seminar in Heterogeneous Catalysis: "Carbon only catalysis" Dr. Dangsheng Su, Fritz Haber Institute of the Max Planck Society, 20.06.2008

KinCat Seminar in Heterogeneous Catalysis: "Pressure Swing Reforming" Dr. Frank Hershkowitz, ExxonMobil, Clinton, NJ, USA, 24.09.2008

KinCat Seminar in Heterogeneous Catalysis: "The Nanoscience Revolution in Catalysis Science" Professor Gabor A. Somorjai, Department of Chemistry, University of California, Berkeley, USA 06.11.2008

International Seminar: Recent Advances in Fibrillar Nanocellulose Research, 40 participants Arranged by PFI, SINTEF and NTNU 12. - 13.11.2008

Guest lectures:

Ph.D. Stephen A. Bedell, The Dow Chemical Co *"Aqueous Amines as Reactive Solvents for Mercaptan Removal"* 22.02.2008

Professor James Rawlings, Department of Chemical Engineering, University of Wisconsin

"Unreachable Setpoints and Economic Optimization in Model Predictive Control (MPC)" 02.04.2008

Professor Rakesh Agrawal, School of Chemical Engineering Purdue University, West Lafayette, IN, USA

"Energy Saving Opportunities in Multicomponent Distillation: Optimum Configuration and Thermal Coupling between Distillation Columns" 15.05.2008

Professor Rolf Findeisen, Max Plank Inst., Magdeburg, Germany

"New approaches for sensitivity analysis and identification in systems biology" 08.10.2008

Dr. Simon Holland, University of the Witwaterrand, Johannesburg, South Africa.

"Reducing CO₂ emmisions for coal to liquid", 24.11.2008

CHAPTER 5: ORGANIZATION - ECONOMY

Organization (also see cover page)

The Head of department is elected for a four-year period (mid 2005 – mid 2009). The scientific staff is divided in to five research groups. In 2008 the former groups Reactor technology and Separation and enviromental technology merged in to The Enviromental engineering and reactor technology group. Each research group has a representative in the management team. The management team has also

Administrative responsibilities of faculty

Faculty Educational Committee (Department representatives)

Faculty Research Committee (Department representatives)

Study Program Chemical Engineering and Biotechnology (Industriell kjemi og bioteknologi) (Department representatives)

Exchange of Norgwegian students taking courses abroad (approval of course program), and approval of course program for visiting exchange students. International Master program

Department Economy

The department has three main sources of income: regular funds from the University, strategic funding from the University and external projects. In terms of external funds, including contributions from the research Council and industry, the situation is very good. However, a recent change in the NT-faculty budgeting principles resulted that the amounts trasferred between faculty and department level has representatives from the PhD's, the students and technical staff. The management team meets every second week and discusses running matters.

In addition to the scientific staff the department has 16 persons in a technical and administrative staff to support teaching and research of all the research groups.

Professor Edd Anders Blekkan, deputy: Associate Professor Jens-Petter Andreassen Professor Hallvard Fjøsne Svendsen, deputy: Professor Hugo Atle Jakobsen Professor Øyvind W. Gregersen

Professor Øyvind W. Gregersen

Associate Professor Jens-Petter Andreassen

changed. The department now gets the regular funds according to several distribution formulas, set by the faculty. The gross regular funds from the University were about 33 million NOK in 2008. These funds are mainly used to pay salaries to the permanent staff. The big increase is due to a new pratice of charging ovehead from and to the external projects. More details are shown in Table 1.

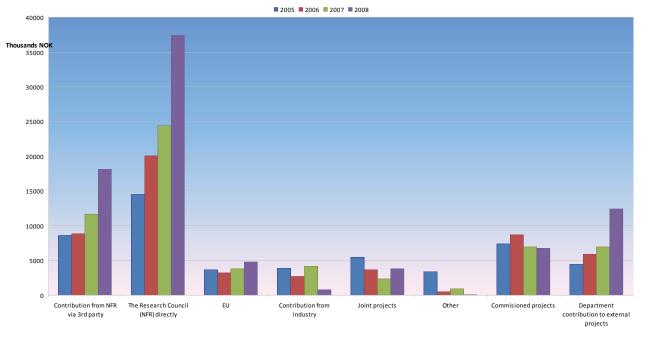
Accounts	2005	2006	2007	2008
Income:				
University funding	22 129 000	24 434 113	26 554 529	33 064 000
Overhead external				
projects	2 153 815	2 390 000	2 846 260	
Sum income	24 282 815	26 824 113	29 402 796	33 064 000
Expences:				
Wages	19 216 318	22 514 217	21 260 668	21 525 000
Investment			2 546 924	147 000
Operating expenses	3 660 005	3 818 963	3 472 929	2 778 000
NTNU contribution				
to ext. projects				9 053 000
Sum expenses	22 876 323	26 333 180	27 280 521	33 504 000
Result	1 406 492	490 933	2 120 268	-439 000

Table 1. Departments income from University and spending.

External funding

In addition to the funding shown in Table 1, the Department has a yearly income from external contributors of approx 72.9 million NOK. Details are shown in Tables 2 and 3. In the end of 2007 and in 2008 we got severalt big projects that have raised the income substatially. Most of the costs are related to salary for PhD candidates. The main contributor to the

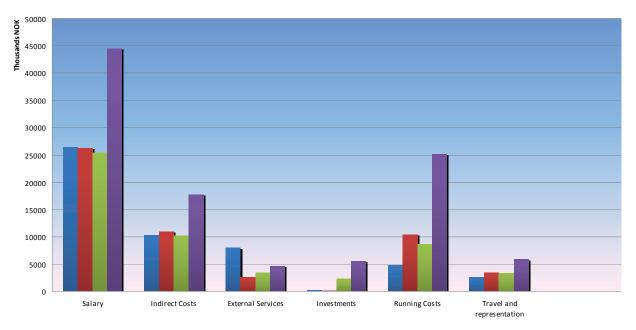
external research activity is The Norwegian Research Council (NFR). Most of these projects are at the Department, but the second largest external source is NFR projects where we contribute as a third party. Industrial contributions and commissioned research has been growing, but the level of funding has now stabilized. The growth in EU-funding is pleasant, and is important because of The University focus on EUfunding that gives extra credit from the Ministry. We have several joint industrial programmes with industry partners from countries in Europe, North-America, South America and Asia. External funding from commissioned research was approx 9 % in 2008, and the customers comes from the mentioned areas and Africa.



Sources of external funding

Use of external funds by category

2005 2006 2007 2008



Strategic funds from University

The third source of income is strategic funding from the University to support research and academic profile. These funds are used for investments in research equipment or PhD or Post doc. positions. In 2008 we did not get any new strategic projects, but concentrated

Net funding from NTNU

From the numbers in Table 1 it seems like the funding from the University has increased heavily in the last

to fill the positions we got the previous year because of the good evaluation we obtained on our application for Center of Excellence where 2 of the applications went to the final round.

few years. However, the departments contributes overhead on external project to the NT faculty, which over the same period has increased from 2686 (2005), 3292 (2006), 3633 (2007), 12658 (2008) and 17800 (2009)(all in thousand NOK). The estimated jump by about 9 millions NOK in 2008 from 2007 to 2008 is because of a recent increase in overhead on salaries

(from 15 % to 40 %.) Subtracting the overhead to the NT faculty, the net funding from the University to the Department over the last few years is 19443 (2005), 19550 (2006), 21764 (2007), 19582 (2008) and estimated 19577 (2009)(all in thousand NOK).

Some pictures of different activities at the Department



Our buildings, from right: K4, K5, Experimental halls and PFI-building (blue)





Eleni, Fredrik, Asbjørn, Magne and Marius



Camilla and Sigurd



Renate, Li, Siv, Øyvind and Heinz



Cabinet minister for Energy, Åslaug Haga, visited the CO₂ absorption rig in April 2008. Here with Hallvard Svendsen



Esther Ochoa-Fernandez received the Exxon Mobil prize for best technical PhD thesis at NTNU in 2008



Researchers night, September 2008. Arne, Kirsti and Nina



All students that work in the labs must attend a safety course

LIST OF TELEPHONE NUMBERS (03.04.2009)

94190	Ahmad, Jamil, PhD candidate	PFI-3402		94114	Lie Jon Arvid, Post doctor	K5-247
50331	Alsvik, Inger Lise	K5-237		50346	Linhart Andreas, PhD candidate	K1-212
95878	Amiri, Asal, PhD candidate	PFI-3405			Luo, Xiao, PhD candidate K1-304	
94209	Andreassen Jens-Petter, Associate Profess	or K5-208		94120	Malthe-Sørenssen Dick, Adjunct Professor	K5-238
	Aronu, Ugochukwu Edwin, PhD candidate	K1-304		93691	Manum Henrik, PhD candidate	K1-326
97018	Bakthiary Hamidreza, PhD candidate	K5-M12		95498	Marchetti, Jorge Mario, Post doctor	K5-156
50318	Barø Tove, Executive Officer	K5-101		94153	Mathisen Torgrim, Higher Executive Officer	K5-101c
92837	Beck Ralf, PhD candidate	K5-146		94073	Mihai, Oana, PhD candidate	K5-M6
94138	Berge, Arvid, Professor Emeritus	K5-M11		94018	Mingot, Andrea, Senior Engineer	K1-309
94157	Blekkan Edd Anders, Professor	K5-429		94032	Moe Størker, Associate Professor	PFI-2108
90638	Borge, Tone, PhD candidate	K5-225		94147		K5-M11
91664		PFI-2101		94193	Moljord, Kjell, Adjunct Professor Muthuswamy, Navaneethan, PhD candidate	
	Borthen Berit, Chief Engineer				-	
94141	Boullosa Eiras Sara, PhD candidate	K5-411		94148	Mørk Preben C., Professor	K5-M11
50326	Bruder, Peter, PhD candidate	K1-213		94022	Nayak, Ameeya Kumar, Post doctor	K5-156
94144	Brun Harry, Engineer	K5-341		94112	Nilsen Tom-Nils, Senior Researcher	K1-310
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93149	Chen De, Professor	K5-407		94111	Noor, Tayyaba, PhD candidate	Kh-251
94163	Ciftja, Arlinda, PhD candidate	K1-123		50924	Opedal, Nils van der Tuuk, PhD candidate	PFI-3408
94159	Dagsgård, Camilla Israelsen	PFI-3407		91559	Panahi, Mehdi, PhD candidate	K1-224
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94110	Dupuy Pablo, PhD candidate	K5-244		98354	Radstake, Paul, PhD candidate	K5-M4
91559	Dwivedi, Deeptanshu, PhD candidate	K1-224		50327	Rafiee, Ahmad, PhD candidate	K1-222
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94120	Erga Olav, Professor Emeritus	K5-237		91686	Rodionova, Galina, PhD candidate	PFI-3209
95878	Fan, Yanru, Post doctor	PFI-3405		94139	Roel Jan Morten, Engineer	Kh-155
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94017	Gawel, Bartlomiej, Researcher	K5-336		94147	Rytter Erling, Adjunct Professor	K5-M11
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50372	Ghadrdan, Maryam, PhD candidate	K1-214		50318	Samseth, Jon, Adjunct Professor	K4-312
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94031	He, Xuezhong, PhD candidate Helle Torbjørn, Professor emeritus	PFI-2109		94124	Skjøndal-Bar, Nadav, Associate Professor	K1-313
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50304	Helmersen Tom, Office Manager			94154	Skogestad Sigurd, Professor	K1-322
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51128	Huang, Fan, PhD candidate	K5-443		91668	Tanase, Mihaela, PhD candidate	PFI-3205
94033	Hägg May-Britt, Professor	K5-204		50346	Tellefsen, Silje Kufaas, PhD candidate	K1-212
91673	lotti, Marco, Post doctor	PFI-3204		94119	Thorsen Gunnar, Professor Emeritus	K5-M8
50537	Janga, Kando Kalifa, PhD candidate	PFI-3207			Trollebø, Anastasia, PhD candidate	K1-304
50327	Jacobsen, Magnus Glosli, PhD candidate	K1-222		50322	Tronstad, Ingvild, PhD candidate	Kh-109
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50331	Karlsen, Cathrine Hval, Seniro Engineer	K5-237		90638	Veiga, Adrian Maroni, PhD candidate	K5-225
94145	Kazi, Saima Sultana, PhD candidate	K5-443		92831	Venvik Hilde, Associate Professor	K5-406
94149	Keleşoĝlu, Serkan, PhD candidate	K5-308		50346	Vevelstad, Solrun, PhD candidate	K1-212
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50540	Lesaint Caterina, Senior Engineer	PFI-3404		50372	Wang, Yuefa, Post doctor	K5-156
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50536	Xhanari, Klodian, PhD candidate	PFI-3206	94073	Zárubová, Šárka, PhD candidate	K5-M6
50537	Yamakawa, Asuka, Researcher	PFI-3207	94155	Zhao Tiejun, Post doctor	K5-432
93146	Yang, Jia, PhD candidate	K5-411	95879	Zhu Zhengjie, PhD candidate	K5-245
50326	Yelchuru, Ramprasad, PhD candidate	K1-213	94018	Øye Asbjørn, Chief Engineer	K1-309
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